

# Cloning, expression, and characterization of a nitric oxide synthase protein from *Deinococcus radiodurans*

Subrata Adak\*<sup>†</sup>, Alexandrine M. Bilwes<sup>‡</sup>, Koustubh Panda\*, David Hosfield<sup>§</sup>, Kulwant S. Aulak\*, John F. McDonald\*, John A. Tainer<sup>§</sup>, Elizabeth D. Getzoff<sup>§</sup>, Brian R. Crane<sup>‡</sup>, and Dennis J. Stuehr\*<sup>†</sup>

\*Department of Immunology, Cleveland Clinic, Cleveland, OH 44195; <sup>†</sup>Department of Chemistry and Chemical Biology, Cornell University, Ithaca, NY 14850; and <sup>§</sup>Department of Molecular Biology, The Scripps Research Institute, La Jolla, CA 92037

Edited by Robert F. Furchgott, State University of New York Downstate Medical Center, Hewlett, NY, and approved October 26, 2001 (received for review September 5, 2001)

We cloned, expressed, and characterized a heme protein from *Deinococcus radiodurans* (*D. radiodurans* NO synthase, deiNOS) whose sequence is 34% identical to the oxygenase domain of mammalian NO synthases (NOSoxys). deiNOS was dimeric, bound substrate Arg and cofactor tetrahydrobiopterin, and had a normal heme environment, despite its missing N-terminal structures that in NOSoxy bind Zn<sup>2+</sup> and tetrahydrobiopterin and help form an active dimer. The deiNOS heme accepted electrons from a mammalian NOS reductase and generated NO at rates that met or exceeded NOSoxy. Activity required bound tetrahydrobiopterin or tetrahydrofolate and was linked to formation and disappearance of a typical heme-dioxy catalytic intermediate. Thus, bacterial NOS-like proteins are surprisingly similar to mammalian NOSs and broaden our perspective of NO biochemistry and function.

Genes coding for NO synthases (NOSs, EC 4.14.23) are present throughout the plant and animal kingdom. NOS activities are present in plants and lower eukaryotes (1–5). Their primary structures and activities are strikingly similar to the mammalian NOSs, suggesting that NO has been important throughout evolution. All eukaryotic NOSs catalyze the NADPH- and O<sub>2</sub>-dependent oxidation of L-arginine (Arg) to citrulline and NO, with *N*-hydroxy-L-arginine (NOHA) formed as an enzyme-bound intermediate (6). Structurally, all animal NOSs are bi-domain proteins containing an N-terminal oxygenase domain (NOSoxy) that binds protoporphyrin IX (heme), 6R-tetrahydrobiopterin (H<sub>4</sub>B), and Arg and is linked to a C-terminal flavoprotein domain (NOS reductase domain, NOSred) by a central calmodulin (CaM) binding sequence (7, 8). NOSred bears strong sequence and functional similarity to NADPH-cytochrome P450 reductase and related electron transfer flavoproteins (9, 10), and function to transfer NADPH-derived electrons to the ferric heme for O<sub>2</sub> activation during NO synthesis. In contrast, NOSoxy and cytochromes P450 have completely different primary, secondary, and tertiary structures, even though both enzyme families use a thiolate-ligated heme for O<sub>2</sub> activation (11, 12). Moreover, unlike P450s, NOSoxy must dimerize to become active (13, 14). Dimerization produces functional binding sites for Arg and H<sub>4</sub>B and sequesters the heme catalytic center from solvent. These distinguishing features imply that NOSs evolved separately from other heme-thiolate enzymes and can so provide unique perspectives on their structure-function relationships.

Although oxidative (nitrification) or reductive (denitrification) pathways for prokaryote NO biosynthesis are well established (15), there have been few reports on NOS-like proteins in bacteria (16, 17). To date no prokaryotic NOS proteins have been completely sequenced, purified, or cloned. However, some have been shown to have nitrite-forming activity that depended on Arg, NADPH, and H<sub>4</sub>B. More recent genome sequencing of *Bacillus subtilis*, *Deinococcus radiodurans*, and other bacteria (18–20) confirm that a subset contain ORFs that code for proteins homologous to NOSoxy. To better understand structure-function aspects of NOS proteins, we cloned, expressed, and

characterized the NOSoxy-like protein in *D. radiodurans* (deiNOS). Here we determine physical and catalytic features of deiNOS and compare them to mammalian NOSoxy. The results reveal striking similarities and differences regarding primary structure, heme environment, utilization of H<sub>4</sub>B or tetrahydrofolate (THF), electron transfer interactions, formation and disappearance of a heme FeIO<sub>2</sub> catalytic intermediate, and activity. We also show that NO synthesis is possible for deiNOS. Together these findings give an important perspective on NOS biochemistry and biological function.

## Materials and Methods

**Materials.** All reagents and materials were obtained from Sigma or sources previously reported (9–11, 21).

**Molecular Biology.** The NOS gene of *D. radiodurans* (ATCC strain number 13939) was amplified by PCR from genomic DNA. PCR primers generated a *Nde*I site before the 5' start codon and a *Bam*HI site after the 3' stop codon, and the amplified fragment was cloned into a pET15B expression vector. The reductase domain of neuronal NOS (nNOS) was created by using PCR to introduce a *Nde*I site, start codon at Met-695, and a *Spe*I site at nucleotide 3040 of nNOS. This PCR product was cloned into the *Nde*I/*Spe*I-digested nNOS pCWori. The cDNA constructs were confirmed by DNA sequencing. Protein expression was carried out in *Escherichia coli* strain BL21 (DE3).

**Expression and Purification of deiNOS, nNOSoxy, and nNOSred.** They were purified by using on Ni<sup>2+</sup>-NTA resin for nNOSoxy and deiNOS and 2',5'-ADP-Sepharose for nNOSred (9, 22). The ferrous-CO adduct absorbing at 444 nm was used to quantitate heme protein content by using an extinction coefficient of 74 mM<sup>-1</sup>·cm<sup>-1</sup> (A<sub>444</sub>–A<sub>500</sub>). The nNOSred concentration was calculated by using an extinction coefficient of 18 mM<sup>-1</sup>·cm<sup>-1</sup> (A<sub>457</sub>–A<sub>800</sub>).

**Arg Binding.** Arg binding affinity was studied at 25°C by perturbation difference spectroscopy using 10 mM imidazole using methods as described (23).

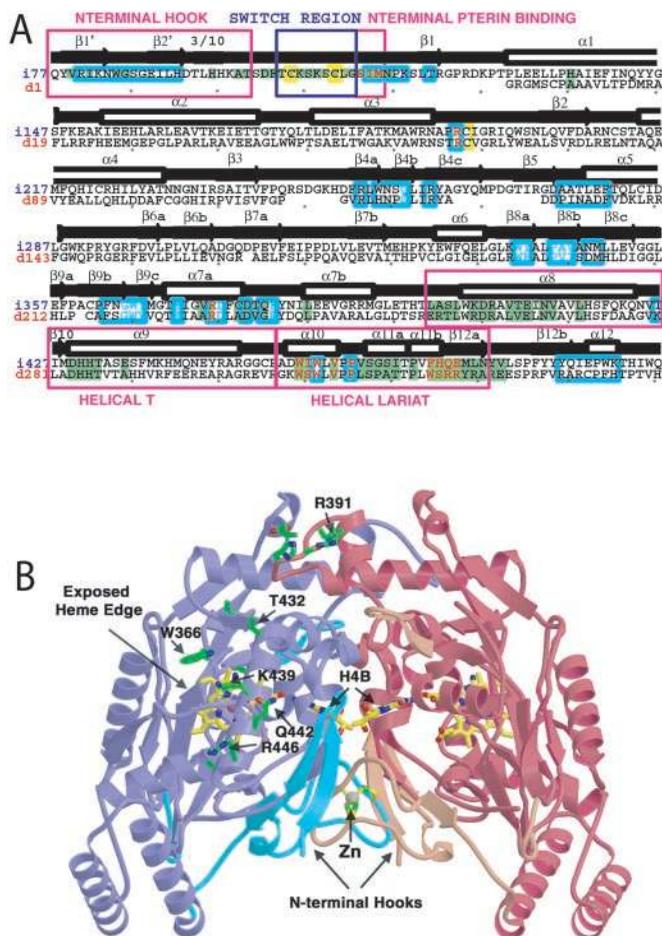
**NO Synthesis, NADPH Oxidation, Nitrite Formation, and Citrulline Production.** The steady-state activities in the reconstituted system containing deiNOS (or nNOSoxy) and nNOSred in a 1:1.5 molar ratio or in the heterodimer system containing deiNOS (or

This paper was submitted directly (Track II) to the PNAS office.

Abbreviations: NOS, NO synthase; NOSoxy, NOS oxygenase domain; NOSred, NOS reductase domain; deiNOS, *D. radiodurans* NOS; nNOS, neuronal NOS; iNOS, inducible NOS; CaM, calmodulin; H<sub>4</sub>B, 6R-tetrahydrobiopterin; NOHA, *N*-hydroxy-L-arginine; THF, tetrahydrofolate.

<sup>†</sup>To whom reprint requests should be addressed at: Immunology NB-3, Cleveland Clinic, 9500 Euclid Avenue, Cleveland, OH 44195. E-mail: adaks@ccf.org or stuehrd@ccf.org.

The publication costs of this article were defrayed in part by page charge payment. This article must therefore be hereby marked "advertisement" in accordance with 18 U.S.C. §1734 solely to indicate this fact.



**Fig. 1.** (A) Aligned iNOSox (above) and deiNOS (below) sequences with mapped residue function, secondary structure, and contributions to the dimer interface determined from iNOSox structures. NOS sequences are color-coded to highlight zinc ligands Cys-104 and Cys-109 and proximal heme ligand Cys-194 (yellow background), Arg-binding residues (cyan letters), H<sub>4</sub>B-binding residues (red letters), and residues that form the active-center channel leading to the heme (cyan boxed). Dimer interface residues that contribute at least 5 Å<sup>2</sup> of buried surface area in the unswapped state of the inducible NOSox domain (34) are shown with green background. Above, black arrows show β-strands, white boxes show α-helices. Key sequence stretches involved in forming the dimer interface and cofactor binding sites are boxed in magenta and denoted, N-terminal hook, switch region (zinc loop), N-terminal pterin binding, helical T, and helical liarat. Seven additional C-terminal deiNOS residues (TGHAPTQ) do not have analogs in the NOSox structures and are thus not shown. (B) Ribbon representation of the unwrapped iNOSox dimer (34) highlighting the structural elements absent in deiNOS and surface residues conserved among NOSs. Most of the NOSox core (purple and red subunits) is conserved by deiNOS with the exception of the N-terminal hook, switch region, N-terminal binding pterin segments, and two peripheral loops (cyan and orange). Notably, the intermolecular zinc binding site (gray, bottom center) is not found in deiNOS, although regions that bind heme (yellow bonds) and H<sub>4</sub>B (yellow, center, on edge) are conserved. Residues that form a conserved surface patch above the active center channel and surrounding an exposed heme edge in mammalian NOSoxs (34) and are also conserved in deiNOS are highlighted (green side chains) and numbered.

nNOSox) monomer and G671A nNOS full-length monomer (5:1) were both determined at 25°C by using spectroscopic and HPLC fluorometric assays as described in detail (8, 24, 25).

**Measurement of Apparent  $k_m$  for H<sub>4</sub>B or THF.** Apparent  $k_m$  values were determined by double reciprocal analysis of the NADPH-dependent nitrite formation against various concentrations of

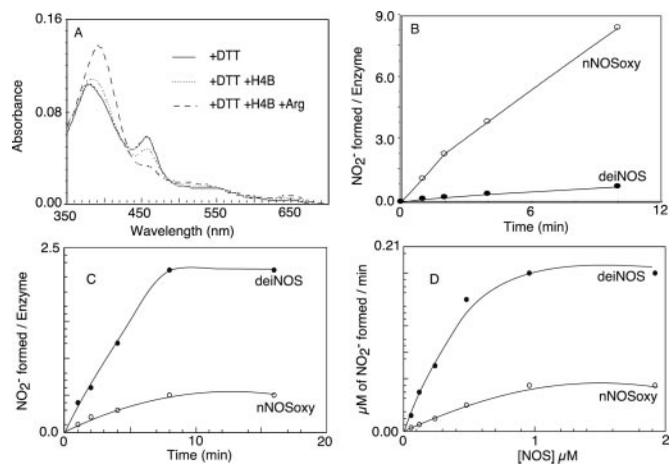
H<sub>4</sub>B or THF in a reconstitution system containing nNOSred and either deiNOS or nNOSox.

**Heme Reduction.** Optical spectra and kinetics were recorded on a Hitachi 3010 UV-visible spectrophotometer at 25°C. Reduction of heme groups was followed anaerobically in a reconstitution system containing deiNOS (or nNOSox) plus nNOSred in a 1:1.5 molar ratio in the presence or absence of CaM under CO gas as described (21). The heme reduction rate was monitored at 444 nm. Anaerobic reaction solutions contained ferric deiNOS or nNOSox (2.8 μM) and nNOSred (4.2 μM), 40 mM 4-(2-hydroxyethyl)-1-piperazine-propanesulfonic acid (EPPS) buffer (pH 7.6), 300 mM NaCl, 0.9 mM EDTA, 1 mM DTT, 1 mM Arg, 20 μM H<sub>4</sub>B, with or without 6 μM CaM plus 1 mM Ca<sup>2+</sup>. An aliquot of NADPH solution was added to give 50 μM to initiate heme reduction.

**Optical Spectroscopy of Heme FeIIO<sub>2</sub> Formation and Decay.** Oxygen binding spectra were recorded in a stopped-flow instrument equipped with a rapid scanning diode array device (Hi-Tech MG-6000, Salisbury, UK). Rapid scanning experiments involved mixing anaerobic solutions containing dithionite-reduced deiNOS or nNOSox, 40 mM EPPS buffer (pH 7.6), 0.5 mM DTT, 300 μM NaCl, and 1 mM Arg with air-saturated buffer solutions at 10°C in the presence or absence of 1 mM H<sub>4</sub>B. Formation and decay of the enzyme heme complexes were followed at 410 nm (407 nm for deiNOS) or 440 nm (22). Signal-to-noise ratios were improved by averaging 10 individual traces. The time courses of FeIIO<sub>2</sub> complex formation and decay were best-fit to a single exponential equation with a nonlinear least-squares method as described (22).

## Results

**Primary Structure Analysis.** Cloning identified a 1-kb cDNA for deiNOS, coding for a 359-aa protein that has 34% identity and 52% conservation with rat nNOSox, and 34% identity and 50% conservation with mouse inducible NOSox (iNOSox). Similarities between primary sequences of deiNOS and mouse iNOSox are striking (Fig. 1A). Structural elements that make up the iNOS catalytic core are well conserved in deiNOS. This includes residues that contact the heme, bind the pteridine ring of H<sub>4</sub>B, and position substrate Arg above the heme. Residues



**Fig. 2.** (A) Spectrum of DTT-bound deiNOS and displacement of DTT upon H<sub>4</sub>B and Arg binding after incubating for 1 h at 25°C. (B) Kinetics of H<sub>2</sub>O<sub>2</sub>-dependent NOHA oxidation by nNOSox and deiNOS at 25°C was measured as described in *Materials and Methods*. (C and D) Time-dependent and enzyme concentration-dependent nitrite formation from NOHA by nNOSox or deiNOS in a reconstitution assay with nNOSred at 25°C in the presence of H<sub>4</sub>B and CaM and 1 mM NADPH.

**Table 1. UV-visible spectral properties of deiNOS and nNOSoxy with Arg or various heme ligands**

Enzyme-ligand complex	deiNOS, nm		nNOSoxy, nm	
	Soret	Visible	Soret	Visible
Fe <sup>3+</sup> -Arg	393	650	393	650
Fe <sup>3+</sup> -imidazole	427	550	427	550
Fe <sup>3+</sup> -DTT	380, 460	650	380, 460	650
Fe <sup>3+</sup> -NO	440	549, 589	440	549, 589
Fe <sup>3+</sup> -Arg	410	560	412	560
Fe <sup>2+</sup> -NO	436	560	436	560
Fe <sup>2+</sup> -CO	444	540	444	540
Fe <sup>2+</sup> -O <sub>2</sub>	427	560	427	560

that participate in the dimer interface or substrate binding channel are somewhat less conserved between deiNOS and iNOS. Sequence analysis predicts that the overall fold of deiNOS is very similar to iNOSoxy, and threading algorithms predict that the entire deiNOS sequence is highly compatible with structures of mammalian NOSoxy proteins. [The LOOPP threading algorithm (J. Mellar and R. Elber, Cornell Theory Center) indicated with high confidence that the deiNOS primary sequence adopts a topology similar to that of the mammalian NOSoxy dimers. The highest match (score = 15.3, global  $Z = 3.6$ , local  $Z = 3.6$ ) occurred between the deiNOS primary sequence and the structure of mouse delta 114 iNOSoxy subunit (Protein Data Bank code 2nos).] However, a notable difference is that deiNOS lacks an extended portion of N-terminal sequence found in the mammalian enzymes. In eukaryotic NOSs this region codes for a N-terminal hook and metal binding site and contains residues that participate in forming the dimer interface and binding the dihydroxypropyl side chain of H<sub>4</sub>B (Fig. 1). Two smaller deletions downstream in the deiNOS sequence correspond to surface-exposed loops in the eukaryotic NOS structure (Fig. 1B). Thus, deiNOS likely has a topology similar to eukaryotic NOSs and contains residues likely to generate functional Arg and H<sub>4</sub>B binding sites.

**Physical and Spectral Characteristics of deiNOS.** Recombinant deiNOS migrated in a denaturing SDS/PAGE gel at a molecular mass of  $\approx 40$  kDa, identical to its cDNA calculated molecular mass. Its migration in low-temperature PAGE and column filtration both indicated that deiNOS was predominantly dimeric in its native form (data not shown). Spectral changes obtained upon Arg or H<sub>4</sub>B binding are shown in Fig. 2A. Ferric deiNOS showed characteristic absorbance maxima at 460 and 380 nm, indicating its heme bound DTT to form a bis-thiolate species that is indistinguishable from nNOSoxy (24). Adding Arg caused the Soret maximum to shift from 460 to 395 nm, indicating Arg displaced DTT and caused a

shift toward five-coordinate high-spin heme. Adding H<sub>4</sub>B caused a similar spectral transition and generated a broad Soret peak at 400 nm. The data indicate that these molecules bind to deiNOS in a similar manner to eukaryotic NOSs.

We next monitored the ability of Arg to displace heme-bound imidazole in deiNOS and nNOSoxy to determine and compare their Arg binding affinity. Upon addition of Arg there was a concentration-dependent spectral shift that indicated Arg could achieve a complete displacement of heme-bound imidazole in both proteins (data not shown). The apparent  $k_d$  value for Arg in the presence of 10 mM imidazole and 20  $\mu$ M H<sub>4</sub>B was derived by double reciprocal analysis and was  $97 \pm 10 \mu$ M and  $55 \pm 4 \mu$ M in deiNOS and nNOSoxy, respectively. We conclude that their Arg binding affinities are similar.

Spectra of ferric or ferrous deiNOS with small heme ligands like NO, CO, and O<sub>2</sub> also were obtained, and their features are compared with nNOSoxy in Table 1. In general, spectra of deiNOS complexed with these ligands were identical to those previously reported for mammalian NOSs. This similarity extended to deiNOS forming stable ferrous heme-CO or heme-NO complexes in the presence of Arg (26). The spectral data suggest that deiNOS has a similar, if not identical, heme environment as nNOSoxy.

**H<sub>2</sub>O<sub>2</sub>-Supported NOHA Oxidation.** We next examined whether deiNOS would oxidize NOHA to nitrite in a H<sub>2</sub>O<sub>2</sub>-driven reaction. H<sub>2</sub>O<sub>2</sub> binds to the NOS ferric heme to form reactive heme-oxy species that can react with Arg or NOHA in the presence or absence of H<sub>4</sub>B (25). In this reaction the heme need not acquire electrons from the NOSred or H<sub>4</sub>B. In Fig. 2B, the turnover number for deiNOS was 0.7 nitrite min<sup>-1</sup>,  $\approx 12$ -fold slower than nNOSoxy tested in the same assay. Including 0.1 mM H<sub>4</sub>B in the assay failed to stimulate nitrite formation by deiNOS but stimulated nNOSoxy activity about 3-fold. This finding suggests that the deiNOS heme may have lower affinity toward H<sub>2</sub>O<sub>2</sub> or may generate a heme-oxy species that is less reactive toward NOHA than nNOSoxy.

**Catalytic Activity in Two Reconstituted Systems.** NO synthesis by animal NOSs involves electron transfer between reductase and oxygenase domains in a NOS dimer. Because deiNOS has no attached reductase domain, its heme can receive electrons only from an externally added reductase protein. We therefore examined its catalytic ability in two types of reconstitution systems. One involved mixing deiNOS with purified nNOSred that contained a functional CaM binding site, whereas the other involved generation of a heterodimer comprised of one deiNOS oxygenase subunit and one nNOS full-length subunit. The nNOS subunit contained a G671A mutation to prevent its forming a homodimer (24). In similar heterodimers of mammalian NOSs, functional electron transfer occurs only between the reductase domain of the full-length subunit and the surrogate oxygenase

**Table 2. Catalytic activities of deiNOS versus nNOSoxy when supported by nNOSred**

Reaction system	NO <sub>2</sub> <sup>-</sup> production, 10 <sup>-2</sup> min <sup>-1</sup>		Citrulline production, 10 <sup>-2</sup> min <sup>-1</sup>		NO synthesis +CaM
	-CaM	+CaM	-CaM	+CaM	
nNOSoxy + Arg	ND	ND	ND	ND	ND
nNOSoxy + NOHA	0.6 $\pm$ 0.1	1.2 $\pm$ 0.1	0.9 $\pm$ 0.1	1.3 $\pm$ 0.2	ND
deiNOS + Arg	ND	ND	ND	ND	ND
deiNOS + NOHA	1 $\pm$ 0.2	1.5 $\pm$ 0.2	1 $\pm$ 0.2	1.5 $\pm$ 0.2	ND
nNOSoxy + Arg + H <sub>4</sub> B	1.2 $\pm$ 0.1	1.2 $\pm$ 0.2	1.2 $\pm$ 0.1	1.6 $\pm$ 0.2	ND
nNOSoxy + NOHA + H <sub>4</sub> B	6.2 $\pm$ 1	10 $\pm$ 1	8.7 $\pm$ 1	13 $\pm$ 1	ND
deiNOS + Arg + H <sub>4</sub> B	1.2 $\pm$ 0.1	1.9 $\pm$ 0.1	1.2 $\pm$ 0.1	2.1 $\pm$ 0.1	ND
deiNOS + NOHA + H <sub>4</sub> B	10 $\pm$ 1	31 $\pm$ 3	10.7 $\pm$ 1	37 $\pm$ 3	ND

Values represent the mean and SD for three measurements. ND means not detectable.

**Table 3. Catalytic activities of deiNOS and nNOSoxy heterodimers**

Reaction system	NO <sub>2</sub> <sup>-</sup> production, 10 <sup>-2</sup> min <sup>-1</sup>		Citrulline production, 10 <sup>-2</sup> min <sup>-1</sup>		NO synthesis + CaM
	-CaM	+CaM	-CaM	+CaM	
G671A nNOS/nNOSoxy	ND	160 ± 4	ND	320 ± 3	ND
G671A nNOS/deiNOS	ND	210 ± 20	ND	320 ± 40	ND
G671A nNOS/nNOSoxy + H <sub>4</sub> B	ND	1,010 ± 40	120 ± 10	1,200 ± 60	10.4 ± 0.3
G671A nNOS/deiNOS + H <sub>4</sub> B	ND	820 ± 20	110 ± 10	1,000 ± 100	8.1 ± 0.4

ND, not detectable.

domain (13, 24, 27). We also repeated all reconstitution reactions by using nNOSoxy for direct comparison.

Our initial work revealed that mixtures of nNOSred plus nNOSoxy or deiNOS could catalyze conversion of either Arg or NOHA to products, but activity was greater using NOHA as substrate, consistent with previous NOS reconstitution systems (28, 29). We chose an assay condition where deiNOS or nNOSoxy dimers were present in a 0.7 molar ratio with added nNOSred. Varying the ratio of deiNOS or nNOSoxy to reductase protein from 1:1.5 to 3:1 had no effect on activity (S.A. and D.J.S., unpublished data). As shown in Fig. 2 C and D, nitrite formation from NOHA was time-dependent and enzyme concentration-dependent in reconstitution reactions that contained H<sub>4</sub>B and either deiNOS or nNOSoxy, with deiNOS being about four times more active than nNOSoxy. Compared with intact nNOS, the amount of NADPH consumed per nitrite generated by deiNOS or nNOSoxy was considerably higher in the reconstitution assay (data not shown), consistent with NADPH depletion occurring within 10 min (Fig. 2C). The nitrite synthesis activities of deiNOS and nNOSoxy in the reconstitution assay were ≈200 times less than intact nNOS (21), consistent with the uncoupled NADPH oxidation.

Results from reconstitution assays that mixed nNOSred with deiNOS or nNOSoxy dimers under various conditions are summarized in Table 2. Nitrite and citrulline formation were undetectable in reactions that contained Arg without H<sub>4</sub>B, but were detected in identical assays that contained NOHA as substrate. Adding H<sub>4</sub>B or CaM was clearly not obligatory. However, in reactions that contained CaM and NOHA, H<sub>4</sub>B stimulated the activity of deiNOS and nNOSoxy by 20-fold and 10-fold, respectively. In comparison to nNOSoxy, deiNOS activities ranged from being equivalent to four times more active in the presence of CaM, H<sub>4</sub>B, and NOHA. Despite their nitrite and citrulline production, none of the assays generated detectable NO in the oxyhemoglobin assay, possibly because of uncoupled NADPH oxidation generating superoxide. Our results imply that deiNOS can accept NADPH-derived electrons from free nNOSred, albeit in an uncoupled manner, to convert Arg or NOHA to citrulline and nitrite in a reaction that is further stimulated by H<sub>4</sub>B. Remarkably, under these circumstances the activity of deiNOS was equivalent or greater than nNOSoxy.

Results from our heterodimer reconstitution experiments are summarized in Table 3. Product formation by both deiNOS and nNOSoxy heterodimer depended on CaM binding, consistent with CaM triggering interdomain electron transfer in the nNOS heterodimer (24, 27). The deiNOS and nNOSoxy heterodimers displayed nearly equivalent activities under all circumstances. Moreover, we detected similar rates of NO formation in the oxyhemoglobin assay for both proteins, demonstrating that deiNOS does make NO under this circumstance.

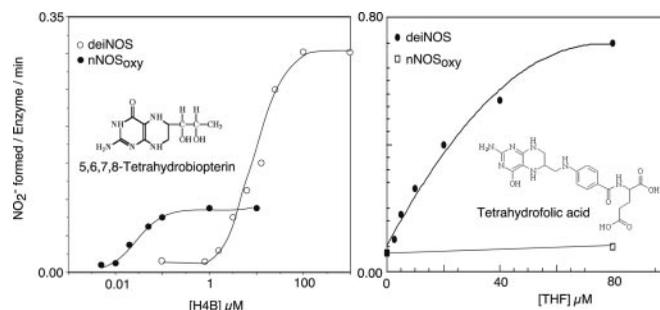
**Affinity Toward H<sub>4</sub>B and THF.** We used the mixed protein reconstitution system to determine an apparent  $K_m$  for H<sub>4</sub>B. Fig. 3 shows that nitrite formation by deiNOS approached  $V_{max}$  conditions near 100  $\mu$ M H<sub>4</sub>B. Reciprocal analysis gave an apparent  $K_m$  for H<sub>4</sub>B of  $10 \pm 2 \mu$ M for deiNOS versus  $30 \pm 10$  nM for

nNOSoxy. Clearly, deiNOS exhibits poorer affinity toward H<sub>4</sub>B than nNOSoxy or other mammalian NOSs, whose apparent  $K_m$  values range between 0.05  $\mu$ M and 1  $\mu$ M.

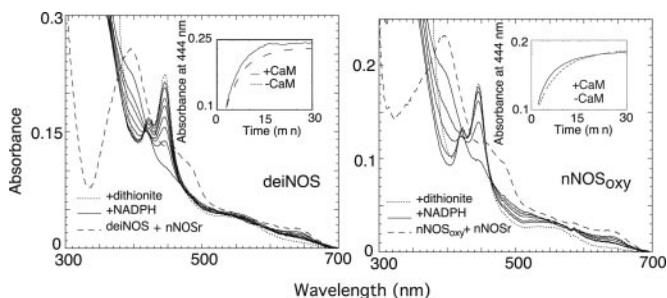
Although *D. radiodurans* does not biosynthesize H<sub>4</sub>B, it does contain THF. Because THF is a tetrahydropteridine like H<sub>4</sub>B, we determined apparent  $K_m$  values for THF in the mixed protein reconstitution assay (Fig. 3). Only deiNOS was found to productively bind THF, with apparent  $K_m$  value of  $20 \pm 5 \mu$ M. The estimated  $V_{max}$  for deiNOS with THF was twice the value obtained with H<sub>4</sub>B. Thus, deiNOS differs from nNOSoxy in using THF in place of H<sub>4</sub>B to support catalysis. This is also true for other bacterial NOS-like proteins (C. S. Raman, personal communication).

**Heme Reduction.** We compared the ability of added nNOSred to catalyze NADPH-dependent heme reduction in deiNOS or nNOSoxy as determined by CO binding under an anaerobic atmosphere. As shown in Fig. 4, about 98% of deiNOS or nNOSoxy heme iron became reduced after NADPH addition to the mixed protein reconstitution system relative to the value obtained by using dithionite, which completely reduces the heme iron independent of nNOSred. Time courses for heme reduction are shown in Fig. 4 *Insets*. Heme reduction was monophasic and slightly faster in the presence of CaM in both cases and was faster in deiNOS compared with nNOSoxy. The apparent rates are listed in Table 4. Faster heme reduction in deiNOS is consistent with its greater activity in this reconstitution assay (see Table 2). However, heme reduction in the reconstituted system was about 1,000 times slower than in intact nNOS when determined under identical conditions (30). This finding is consistent with slower catalytic activity in the reconstituted system.

**Reaction of Ferrous deiNOS with O<sub>2</sub>.** During NO synthesis, formation of a transient FeIIO<sub>2</sub> intermediate is prerequisite for catalysis (22, 31, 32). We used rapid-scanning stopped-flow spectroscopy to examine spectral and kinetic properties of the FeIIO<sub>2</sub> intermediate in deiNOS and to observe whether H<sub>4</sub>B would affect its properties. An anaerobic solution of dithionite-



**Fig. 3.** Affinities and activities of deiNOS and nNOSoxy toward H<sub>4</sub>B and THF. NADPH-dependent nitrite formation was measured as a function of H<sub>4</sub>B (Left) or THF concentration (Right) in a reconstitution assay containing NOHA, CaM, and nNOSred.



**Fig. 4.** Heme iron reduction by nNOSred in deiNOS (Left) versus nNOSoxy (Right) under anaerobic conditions. Spectra were collected before (dashed line) and after addition of 50  $\mu$ M NADPH (solid lines), and then after addition of 100  $\mu$ M dithionite (dotted line). (Insets) The kinetics of heme iron reduction as measured by CO binding in the presence or absence of  $\text{Ca}^{2+}$  and CaM.

reduced deiNOS (or nNOSoxy) containing Arg with or without H<sub>4</sub>B was rapid-mixed with air-saturated buffer at 10°C. Groups of spectra were recorded to follow formation and decay of the FeIO<sub>2</sub> intermediate (Fig. 5). The initial ferrous species had a Soret peak at 407 nm and converted to a transient species with Soret peak at 427 nm in both deiNOS and nNOSoxy. In deiNOS the transient species Soret peak was less well defined and built up to a lesser extent. Both transient species converted to stable ferric enzymes that displayed a Soret peak at 395 nm and visible absorbance band at 650 nm (Fig. 5). Thus, deiNOS formed a FeIO<sub>2</sub> intermediate that is quite similar to nNOSoxy.

Formation and decay kinetics of the FeIO<sub>2</sub> intermediate were determined by monitoring absorbance change at 407 or 440 nm versus time. The direction of absorbance change at these two wavelengths was reversed as expected (see Fig. 5), but otherwise proceeded with identical kinetics (data not shown), as found previously for nNOSoxy (22). Spectral change during the O<sub>2</sub> binding or FeIO<sub>2</sub> decay phases was best described by a single exponential equation in all cases, suggesting both transitions are monophasic. FeIO<sub>2</sub> formation was somewhat slower in deiNOS compared with nNOSoxy at the O<sub>2</sub> concentration used here (Table 4) and was unaffected by added H<sub>4</sub>B in both proteins, consistent with H<sub>4</sub>B not significantly affecting O<sub>2</sub> binding kinetics in animal NOSs (22). FeIO<sub>2</sub> disappearance in H<sub>4</sub>B-free deiNOS was three times faster compared with nNOSoxy. Adding H<sub>4</sub>B to deiNOS increased this rate by about 3-fold, whereas H<sub>4</sub>B addition to nNOSoxy caused a 60-fold acceleration in its FeIO<sub>2</sub> disappearance, as reported (22). Thus, in the presence of H<sub>4</sub>B, FeIO<sub>2</sub> disappearance in deiNOS was three times slower compared with nNOSoxy (Table 4).

## Discussion

The NOS-like protein of *D. radiodurans* provides unique perspective on NOS structure-function. For example, deiNOS was dimeric and catalytically active despite its missing an N-terminal segment that is conserved among most animal NOSs. This segment contains a N-terminal hook that helps form the dimer interface, two cysteine residues that form a metal ion binding site ( $\text{Zn}^{2+}$ ) at the dimer interface, and several residues that bind the

dihydroxypropyl side chain of H<sub>4</sub>B (refs. 11, 12, 33, and 34; see Fig. 1). Apparently, these elements are not essential for deiNOS dimerization, H<sub>4</sub>B and Arg binding, proper heme complex formation with NO, CO, or O<sub>2</sub>, or electron import from a separate reductase domain for catalysis. This finding differs from how loss of these N-terminal elements affect certain animal NOSs. In iNOSoxy, deletion of the first 117 aa removes its N-terminal hook, metal binding site, and residues interacting with the H<sub>4</sub>B side chain. This deletion generated a thiol-ligated heme protein that was monomeric and catalytically inactive and had no capacity to bind H<sub>4</sub>B or Arg or form a homodimer (35). In the case of endothelial NOS, deletion of the first 105 residues also removed the three N-terminal elements, but did not completely abolish dimerization, substrate and H<sub>4</sub>B binding, and catalysis (36). Regarding nNOS, two alternatively spliced variants that are missing the N-terminal hook and metal binding site (37), or missing both of these elements plus residues that bind the H<sub>4</sub>B side chain (38), displayed normal or minimal activity, respectively, when expressed in animal cells. Thus, a picture is emerging where NOS isoforms exhibit a range of dependence on the three structural elements contained in the N terminal, with iNOS at one extreme and deiNOS at the other.

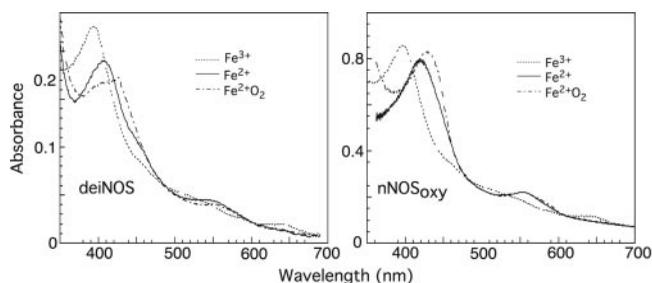
Our current work defines two reconstitution methods (added nNOSred, and heterodimer formation) for studying prokaryotic NOSs regarding catalysis and structure-function. Remarkably, deiNOS oxidized Arg or NOHA when supported by nNOSred and was more active than nNOSoxy in this setting. Its greater activity correlated with a faster rate of heme reduction. DeiNOS must have a competent docking site for the nNOSred to allow electron transfer and catalysis. However, heme reduction was still very slow and was associated with uncoupled NADPH consumption in this particular reconstitution system. Similar results were reported in reconstitution studies with separate iNOS oxygenase and reductase domains (28). This finding is consistent with heme reduction in animal NOSs requiring specific interactions in an intact dimer rather than collisions between free reductase and oxygenase domains in solution. However, because deiNOS has no attached reductase domain, it must normally rely on interactions with a separate redox partner to receive electrons. Our current results suggest that its native redox partner probably differs significantly from nNOSred. Indeed, the genome of *D. radiodurans* codes for no flavoproteins that match cytochrome P450 reductase or nNOSred, although several alternative electron transfer proteins are present (20).

deiNOS also oxidized Arg when it was part of a heterodimer that contained full-length G671A nNOS as the partner subunit. The oxyhemoglobin assay directly demonstrated that deiNOS can generate NO. Remarkably, heterodimers containing deiNOS or nNOSoxy displayed equivalent rates of NO synthesis that were well coupled to their NADPH consumption. Thus, interactions between the deiNOS subunit and the mammalian reductase domain were fully functional in a heterodimer, despite their distant evolutionary relationship and the fact that a 350-aa N-terminal segment is missing in deiNOS. Studies with NOS chimeras have shown that electrons can transfer between domains from different isoforms (i.e., nNOS and endothelial NOS reductase domains transferred electrons at their native rates to either nNOS or endothelial NOS oxygenase domains) (30). Thus, an interaction surface may be

**Table 4. Rates of NADPH-dependent ferric heme reduction and FeIO<sub>2</sub> complex formation and decay in deiNOS and nNOSoxy**

Enzyme	Heme reduction, $10^{-3} \text{ s}^{-1}$		FeIO <sub>2</sub> formation, $\text{s}^{-1}$		FeIO <sub>2</sub> decay, $\text{s}^{-1}$	
	-CaM	+CaM	-H <sub>4</sub> B	+H <sub>4</sub> B	-H <sub>4</sub> B	+H <sub>4</sub> B
nNOSoxy	1.5 ± 0.2	3.0 ± 0.4	200*	163*	0.14*	10*
deiNOS	2.3 ± 0.3	5.7 ± 0.7	72 ± 4	60 ± 6	1.37 ± 0.04	3.9 ± 0.2

\*Values were taken from ref. 22.



**Fig. 5.** Spectrum of a transient  $\text{Fe(II)O}_2$  catalytic intermediate formed in deiNOS (Left) and nNOSoxy (Right). The initial ferrous spectra shown were collected before mixing. Spectra indicating the  $\text{Fe(II)O}_2$  intermediate were collected at  $\approx 7$  ms and 28 ms after mixing for the nNOSoxy and deiNOS reactions, respectively, whereas spectra of the ferric species were collected 3 s after mixing. Other collected spectra were omitted for clarity.

conserved that is independent of the N-terminal elements. The overall conservation of surface residues is predicted to be small between deiNOS and mammalian NOSoxy proteins (for example, 14% identity and 23% homology with mouse iNOSoxy). However, one conserved cluster of surface residues formed from helix  $\alpha 9$  on one subunit and the C-terminal end of helix  $\alpha 7a$  on another (Fig. 1), corresponds to a region identified as a potential interaction site for the mammalian NOSred. This patch, which is adjacent to an exposed heme edge, was implicated in flavoprotein interactions based on conservation of surface residues among eukaryotic NOSs and the electrostatic character of NOSoxy and NOSred surfaces (34). Thus, electron transfer between deiNOS and its native redox partner may rely on similar interactions as those between the mammalian oxygenase and reductase domains. On the other hand, specific surface features may matter little once a foreign oxygenase subunit becomes part of a heterodimer (i.e., electron transfer may be made possible primarily by reductase domain proximity). These possibilities can now be addressed.

The ability of a deiNOS subunit to function so well in the heterodimer could also involve its being “rescued” by a phenomenon known as N-terminal hook swapping. Our previous work with iNOS showed that N-terminal hook swapping occurs in both homodimers and heterodimers (34, 35). Swapping within a heterodimer explains how iNOSoxy subunits can display good activity despite their having N-terminal mutations that debilitate catalysis

in a homodimer. For example, a heterodimer containing an iNOSoxy subunit missing the first 117 aa (and thus similar to deiNOS, see Fig. 1) catalyzed NO synthesis at 50% the rate of wild type, apparently because its partner subunit provided an intact N-terminal hook to the mutant subunit (35). In our current study, the nNOS G671A full-length subunit could have provided its N-terminal hook to deiNOS through a swapping interaction, and in this way made deiNOS more like nNOSoxy regarding structure and catalysis in the heterodimer. On the other hand, it is still possible that deiNOS is constructed so as to not require a N-terminal hook for good catalytic function.

Reduced pteridines bound to deiNOS and participated in its catalysis. This is evidenced by  $\text{H}_4\text{B}$  stimulating deiNOS citrulline formation in the mixed protein reconstitution system and being required for heterodimer NO synthesis. The primary structure of deiNOS is entirely consistent with its behavior toward  $\text{H}_4\text{B}$ . Residues that surround the  $\text{H}_4\text{B}$  ring in mammalian NOSs and position it near the heme are conserved in deiNOS. Conversely, the absence of N-terminal residues that in animal NOSs bind the 6-dihydroxypropyl side chain of  $\text{H}_4\text{B}$  may explain why deiNOS has poorer affinity toward  $\text{H}_4\text{B}$ . The missing N-terminal region probably also allows deiNOS to bind THF, which has a bulkier group in place of the 6-dihydroxypropyl side chain. This characteristic may be biologically important because it suggests how *D. radiodurans* could generate NO despite its lack of  $\text{H}_4\text{B}$ .

In mammalian NOSs,  $\text{H}_4\text{B}$  is implicated as an electron donor (32, 39) and provides an electron to the heme  $\text{Fe(II)O}_2$  intermediate for Arg hydroxylation in a single turnover reaction (31). In this circumstance electron transfer from  $\text{H}_4\text{B}$  speeds disappearance of the  $\text{Fe(II)O}_2$  intermediate (22). Remarkably,  $\text{H}_4\text{B}$  also increased the rate of  $\text{Fe(II)O}_2$  disappearance in deiNOS under identical circumstances. This behavior, together with deiNOS requiring  $\text{H}_4\text{B}$  for NO synthesis and its conserving residues important for  $\text{H}_4\text{B}$  function, strongly suggest that a native reduced pteridine may perform an identical redox function in *D. radiodurans*. Our current work sets the stage for detailed biochemical and biological investigations of bacterial NO synthesis and should provide a deeper understanding of all NOSs.

We thank Kartikeya Pant for help with LOOPP calculations. This work was supported by National Institutes of Health Grants CA53914 (to D.J.S.) and HL58883 (to E.D.G.), and grants from The American Heart Association (to S.A.) and The Camille and Henry Dreyfus Foundation (to B.R.C.).

- Muller, U. (1997) *Prog. Neurobiol.* **51**, 363–381.
- Klesig, D. F., Durner, J., Noad, R., Navarre, D. A., Wendehenne, D., Kumar, D., Zhou, J. M., Shah, J., Zhang, S., Kachroo, P., et al. (2000) *Proc. Natl. Acad. Sci. USA* **97**, 8849–8855.
- Golderer, G., Werner, E. R., Leitner, S., Grobner, P. & Werner-Felmayer, G. (2001) *Genes Dev.* **15**, 1299–1209.
- Ninnemann, H. & Maier, J. (1996) *Photochem. Photobiol.* **64**, 393–398.
- Radomski, M. W., Martin, J. F. & Moncada, S. (1991) *Philos. Trans. R. Soc. London B* **334**, 129–133.
- Griffith, O. W. & Stuehr, D. J. (1995) *Annu. Rev. Physiol.* **57**, 707–736.
- Sheta, E. A., McMillan, K. & Masters, B. S. (1994) *J. Biol. Chem.* **269**, 15147–15153.
- Ghosh, D. K. & Stuehr, D. J. (1995) *Biochemistry* **34**, 801–807.
- Gachhui, R., Presta, A., Bentley, D. F., Abu-Soud, H. M., McArthur, R., Brudvig, G., Ghosh, D. K. & Stuehr, D. J. (1996) *J. Biol. Chem.* **271**, 20594–20602.
- Zhang, J., Martasek, P., Paschke, R., Shea, T., Masters, B. S. & Kim, J. J. (2001) *J. Biol. Chem.* **276**, 37506–37513.
- Crane, B. R., Arvai, A. S., Ghosh, D. K., Wu, C., Getzoff, E. D., Stuehr, D. J. & Tainer, J. A. (1998) *Science* **279**, 2121–2126.
- Fischmann, T. O., Hruza, A., Niu, X. D., Fossetta, J. D., Lunni, C. A., Dolphin, E., Prongay, A. J., Reichert, P., Lundell, D. J., Narula, S. K. & Weber, P. C. (1999) *Nat. Struct. Biol.* **6**, 233–242.
- Siddhanta, U., Presta, A., Fan, B., Wolan, D., Rousseau, D. L. & Stuehr, D. J. (1998) *J. Biol. Chem.* **273**, 18950–18958.
- Klatt, P., Schmidt, K., Lehner, D., Glatter, O., Bachinger, H. P. & Mayer, B. (1995) *EMBO J.* **14**, 3687–3695.
- Goretski, J. & Hollocher, T. C. (1990) *J. Biol. Chem.* **265**, 889–895.
- Chen, Y. & Rosazza, J. P. N. (1994) *Biochem. Biophys. Res. Commun.* **203**, 1251–1258.
- Choi, W. S., Chang, M. S., Han, J. W., Hong, S. Y. & Lee, H. W. (1997) *Biochem. Biophys. Res. Commun.* **237**, 554–558.
- Kunst, F., Ogasawara, N., Moszer, I., Albertini, A. M., Alloni, G. & Azevedo, V. (1997) *Nature (London)* **390**, 249–256.
- White, O., Eisen, J. A., Heidelberg, J. F., Hickey, E. K., Peterson, J. D., Dodson, R. J., Haft, D. H., Gwinn, M. L., Nelson, W. C. & Richardson, D. L. (1999) *Science* **286**, 1571–1577.
- Degtyarenko, K. N. & Kulikova, T. A. (2001) *Biochem. Soc. Trans.* **29**, 139–147.
- Adak, S., Ghosh, S., Abu-Soud, H. M. & Stuehr, D. J. (1999) *J. Biol. Chem.* **274**, 22313–22320.
- Abu-Soud, H. M., Gachhui, R., Raushel, F. M. & Stuehr, D. J. (1997) *J. Biol. Chem.* **272**, 17349–17353.
- Adak, S. & Stuehr, D. J. (2001) *J. Inorg. Biochem.* **83**, 301–308.
- Panda, K., Ghosh, S. & Stuehr, D. J. (2001) *J. Biol. Chem.* **276**, 23349–23356.
- Adak, S., Wang, Q. & Stuehr, D. J. (2000) *J. Biol. Chem.* **275**, 33554–33561.
- Abu-Soud, H. M., Wu, C., Ghosh, D. K. & Stuehr, D. J. (1998) *Biochemistry* **37**, 3777–3786.
- Sagami, I., Daff, S. & Shimizu, T. (2001) *J. Biol. Chem.* **276**, 30036–30042.
- Ghosh, D. K., Abu-Soud, H. M. & Stuehr, D. J. (1995) *Biochemistry* **34**, 11316–11320.
- Chen, P. F., Tsai, A. L., Berka, V. & Wu, K. K. (1996) *J. Biol. Chem.* **271**, 14631–14635.
- Adak, S., Aulak, K. S. & Stuehr, D. J. (2001) *J. Biol. Chem.* **276**, 23246–23252.
- Wei, C. C., Wang, Z. Q., Wang, Q., Meade, A. L., Hemann, C., Hille, R. & Stuehr, D. J. (2001) *J. Biol. Chem.* **276**, 315–319.
- Bec, N., Gorren, A. C., Voelker, C., Mayer, B. & Lange, R. (1998) *J. Biol. Chem.* **273**, 13502–13508.
- Raman, C. S., Li, H., Martasek, P., Kral, V., Masters, B. S. & Poulos, T. L. (1998) *Cell* **95**, 939–950.
- Crane, B. R., Rosenfeld, R. J., Arvai, A. S., Ghosh, D. K., Ghosh, S., Tainer, J. A., Stuehr, D. J. & Getzoff, E. D. (1999) *EMBO J.* **18**, 6271–6281.
- Ghosh, D. K., Crane, B. R., Ghosh, S., Wolan, D., Gachhui, R., Crooks, C., Presta, A., Tainer, J. A., Getzoff, E. D. & Stuehr, D. J. (1999) *EMBO J.* **18**, 6260–6270.
- Rodriguez-Crespo, I., Moenne-Loccoz P., Loehr, T. M. & Ortiz de Montellano, P. R. (1997) *Biochemistry* **36**, 8530–8538.
- Wang, Y., Newton, D. C., Robb, G. B., Kau, C.-L., Miller, T. L., Cheung, A. H., Hall, A. V., VanDamme, S., Wilcox, J. N. & Marsden, P. A. (1999) *Proc. Natl. Acad. Sci. USA* **96**, 12150–12155.
- Brennan, J. E., Xia, H., Chao, D. S., Black, S. M. & Bredt, D. S. (1997) *Dev. Neurosci.* **19**, 224–231.
- Hurshman, A. R., Krebs, C., Edmondson, D. E., Huynh, B. H. & Marletta, M. A. (1999) *Biochemistry* **38**, 15689–15696.